

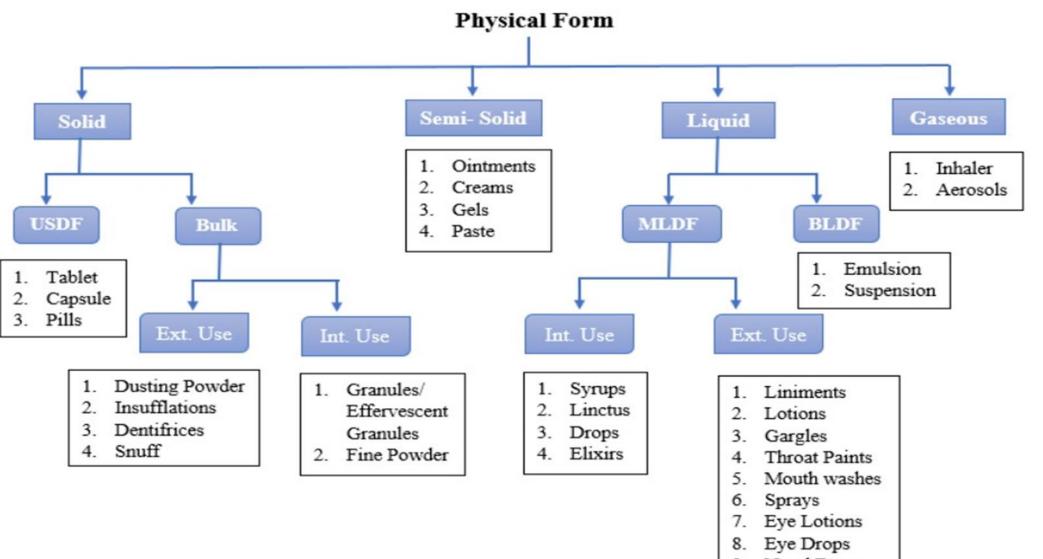


Routes of Drug Administration

By Assistant Lecture Abdulazeez .M.H

Class III





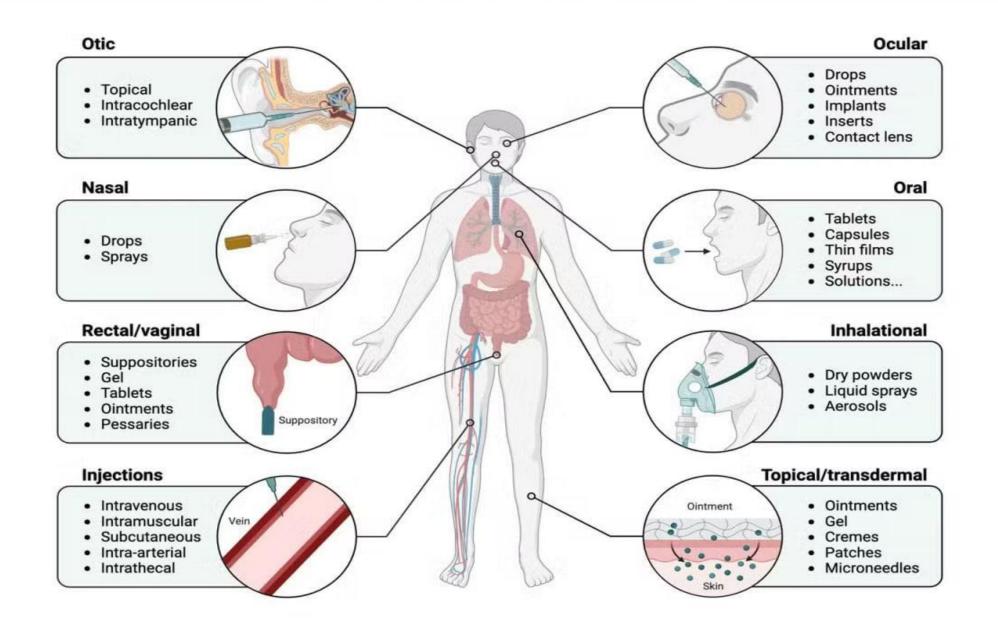
9. Nasal Drops

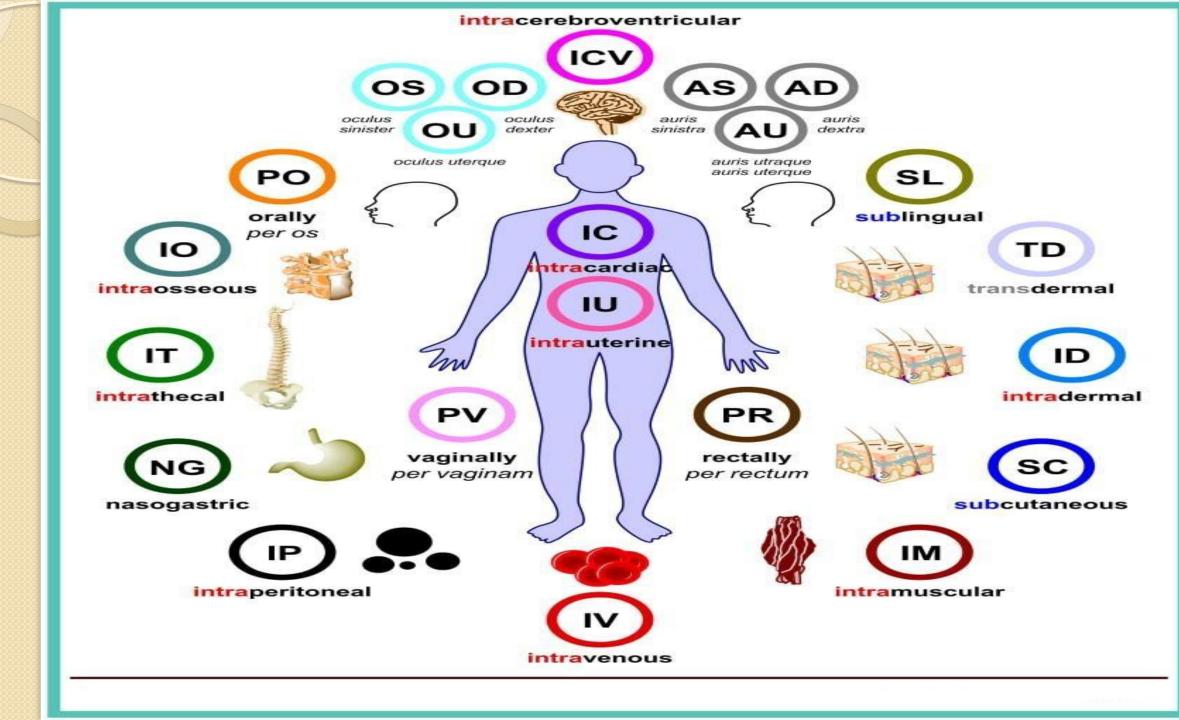
Routes of Administration

 Route of administration is determined by properties of the drug
 e.g. solubility (water or lipid), ionization
 e.g. when we need rapid onset or long treatment regimen.

 At site of action, the drug pharmacological effect, through which the expected therapeutic effect occurs and also possible side effects.

Drug Administration Routes





- The most common, convenient, & economical method
- Swallowed, allowing oral delivery ,
- Placed under the tongue (sublingual)
- Placed between the gums and cheek (buccal)

Advantages

easy self-administered

overcome drug toxicities and/or overdose of oral drugs with antidotes,

However, the pathways of absorption highly complicated, and low gastric pH inactivates some drugs.



Oral Dosage Forms

Delayed-release vs. immediate-release medication

What are pills?

What are tablets?

What are caplets?

Smaller than tablets & coated with a film or gelatin coatingcreates a smoother finish .that makes caplets easier to swallow than tablets

?What are two-piece hard capsules

Composed of two cylindrical shells that fit together that encapsulate powders, granules, beads, .tablets and liquids – or a combination of these items

:Most two-piece hard capsules are made of either gelatin (animal-based or plant-based material .Hydroxypropyl-Methyl Cellulose HPMC

Sublingual (SL) & buccal

Advantages

Easy administration Rapid absorption: 1 – 3 minutes Bypass of gastric acidity Avoidance of first-pass metabolism

- SL tablet
- SL strips
- Multi-purpose SL tablets
- > SL drops
- SL Sprays
- > SL lozenges



Oral inhalation

Rapid delivery of drug across the large surface area of mucous membranes of the respiratory tract and pulmonary epithelium. Drug effects are rapid .. almost like IV bolus.

Rectal Route .. e.g. Suppositories Urethral Route .. e.g. Urethral Bogies Vaginal Route .. e.g. Vaginal Bogies

Parenteral administration

Parenteral Routes

The drug does not pass through the

gastrointestinal tract. It directly reaches to the blood.

May be classified into:

With injections:

e.g. Intravascular, Intramuscular, Subcutaneous

Without injections e.g. Inhalations.

Parenteral administration

Intravascular IV: Absorption phase is bypassed.

Advantages-:

- 1. Precise, accurate and almost immediate onset of action
- 2. Large quantities can be given, fairly pain free
- 3. Can be given to unconscious patients.
- 4. Quick action
- 5. Drugs having unpleasant taste can be given.

Disadvantages-:

- 1. Pain at the site of injection.
- 2. Greater risk of adverse effects

High concentration attained rapidly Risk of embolism

Parenteral administration

Intramuscular (IM)

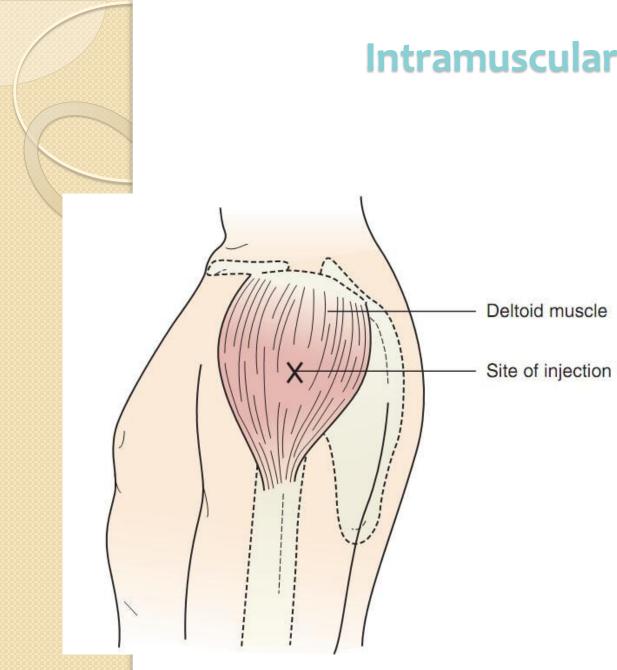
Drug once reaches the muscles absorbs into the blood.

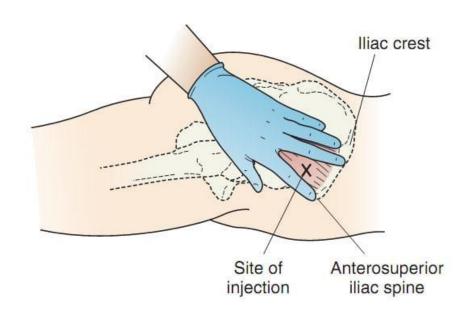
- > Very rapid absorption of drugs in aqueous solution
- > Depot and slow release preparations
- > Pain at injection sites for certain drugs

Subcutaneous (SC)

Drug once reaches to the subcutaneous layer crosses the membrane and absorbed into the blood.

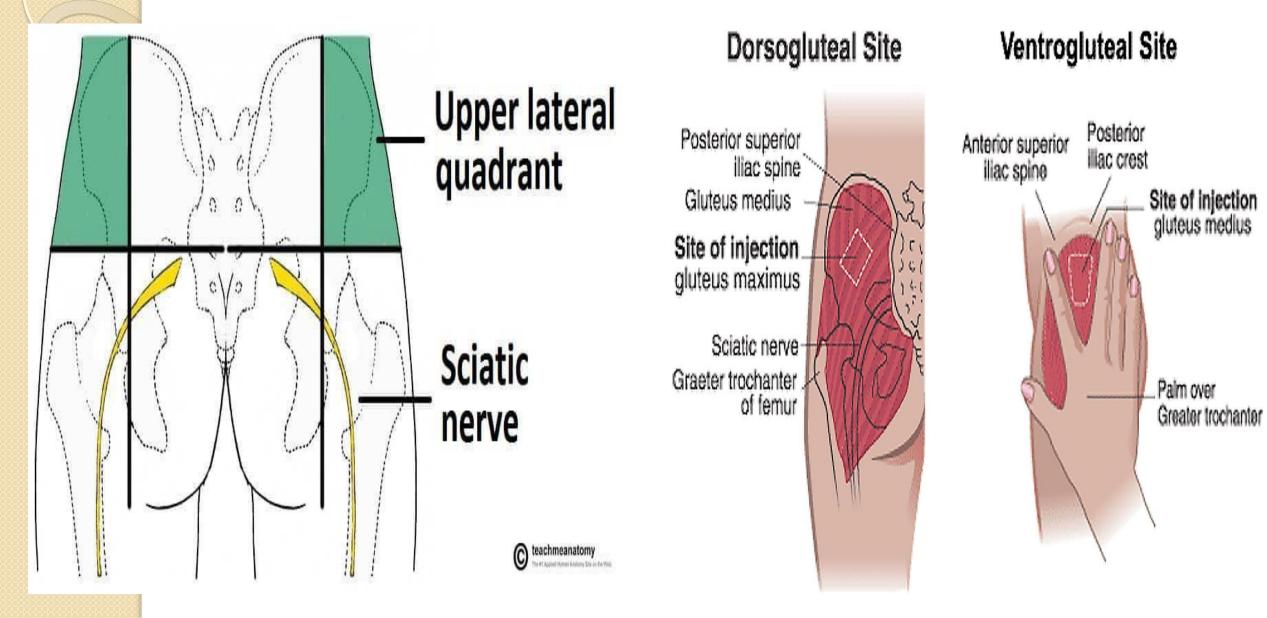
Bypass the GIT.

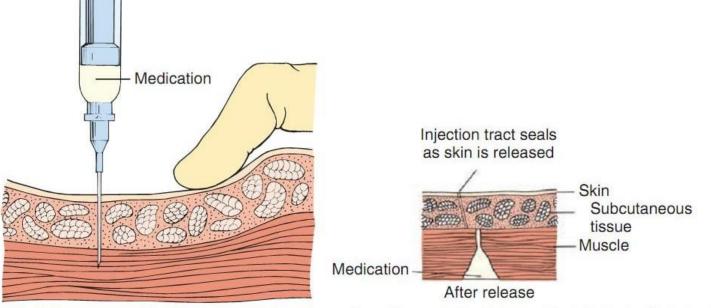




Intramuscular administration

Intramuscular administration

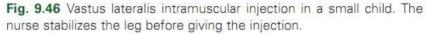




Intramuscular administration

Fig. 9.43 Z-track method for intramuscular injections. (From Perry, A. G., & Potter, P. A. [2014]. *Clinical nurs-ing skills and techniques* [8th ed.]. St. Louis, MO: Mosby.)





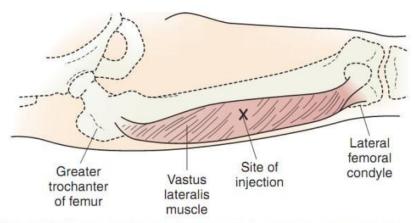


Fig. 9.47 Finding landmarks for a vastus lateralis intramuscular injection.

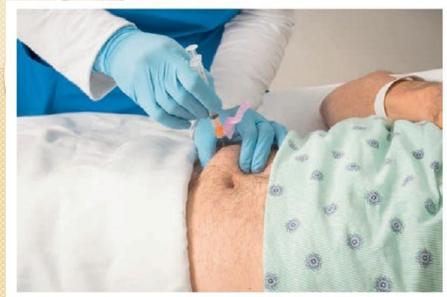


Fig. 9.39 When giving a subcutaneous injection in the abdomen, be sure to choose a site at least 2 inches away from the umbilicus.



Fig. 9.38 Giving a subcutaneous injection at a 90-degree angle.

Subcutaneous administration

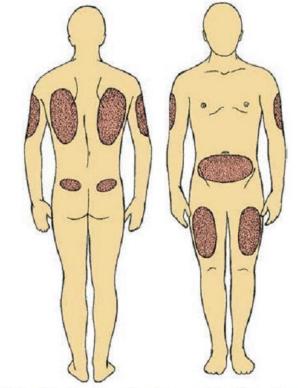
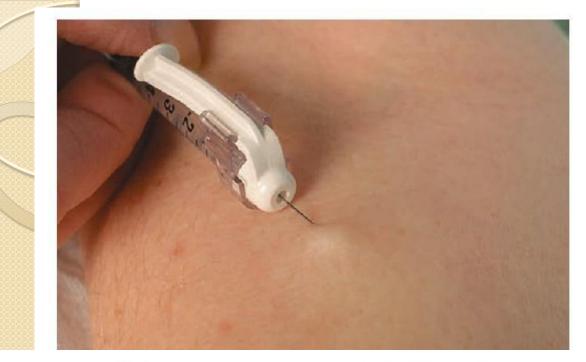
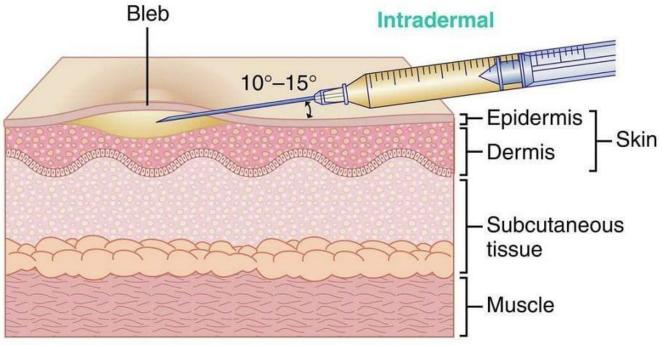
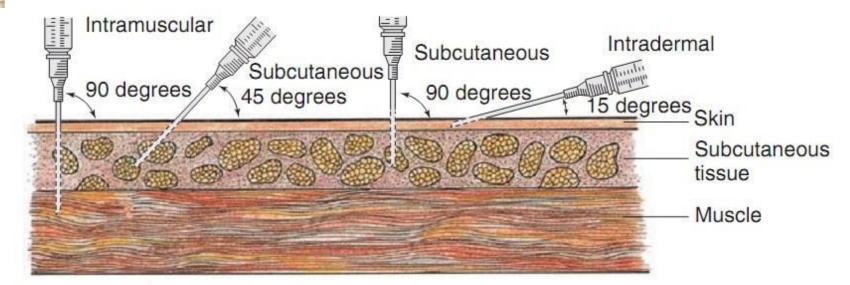


Fig. 9.36 Potential sites for subcutaneous injections. (From Perry, A.

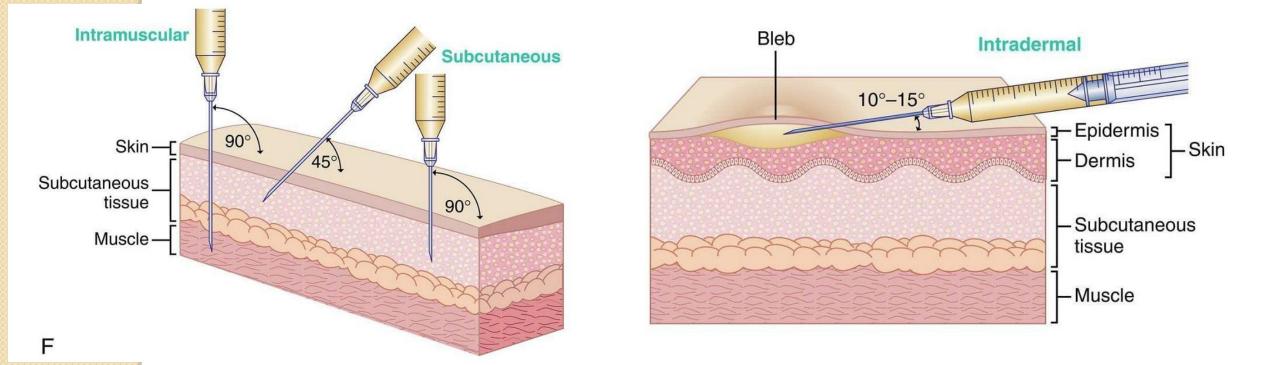


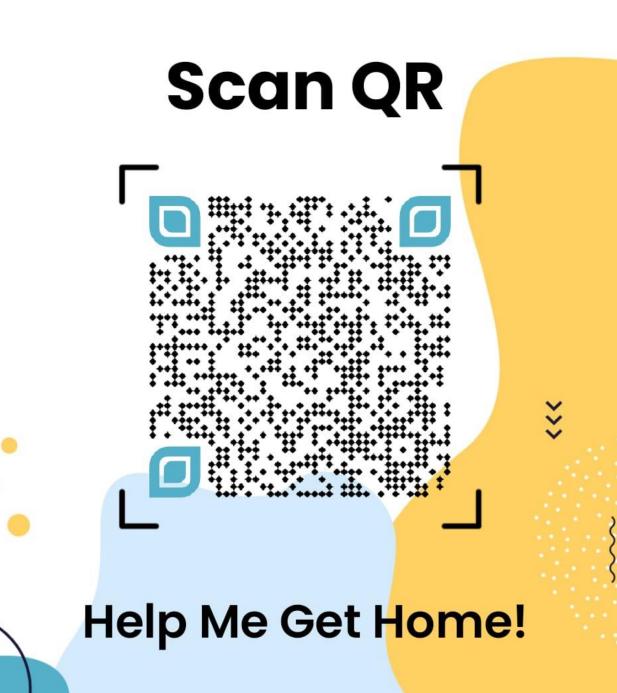
Intradermal administration





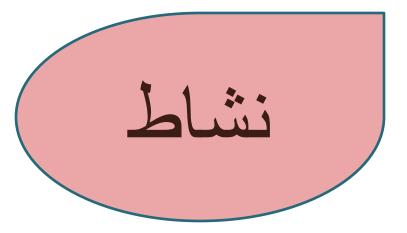
Comparison of angles of needle insertion for injections.





An 18 year old female patient is brought to emergency department due to drug overdose. Which of the following routes of administration is the most desirable for administering the **antidote** for the drug overdose?

A. Intramuscular B. Intravenous C. Oral D. Subcutaneous E. Transdermal



Onset of Action

Onset of action of different routes

- Intravenous 30-60 seconds
- Intraosseous 30-60 seconds
- Inhalation 2-3 minutes
- Sublingual 3-5 minutes
- Intramuscular 10-20 minutes
- Subcutaneous 15-30 minutes
- Rectal 5-30 minutes
- Oral 30-90 minutes
- Topical/transdermal (topical) variable (minutes to hours(



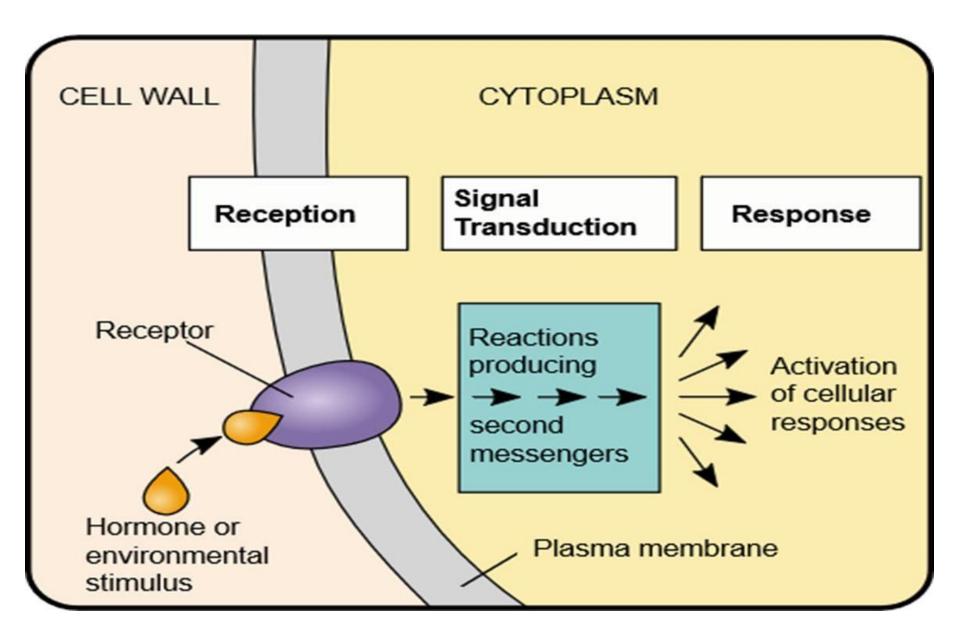
Signal transduction

It is a cellular mechanism converts a stimulus into a response in the cell

Effector molecules or "Second messenger" are part of the cascade of events that translates agonist binding into a cellular response.

Example adrenaline (1st messenger) + β receptor $\rightarrow \uparrow$ activity of adenylyl cyclase $\rightarrow \uparrow$ cAMP {2nd messenger (effector molecule) } \rightarrow response (either beneficial, or harmful \rightarrow adverse effects.(

Signal transduction



https://classroom.google.com/c/NzI4ND g2MDM2NzUw?cjc=wqi6q4r

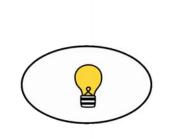


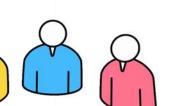
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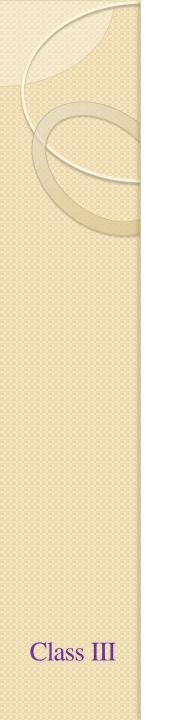
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General Pharmacology

Lab Sessions

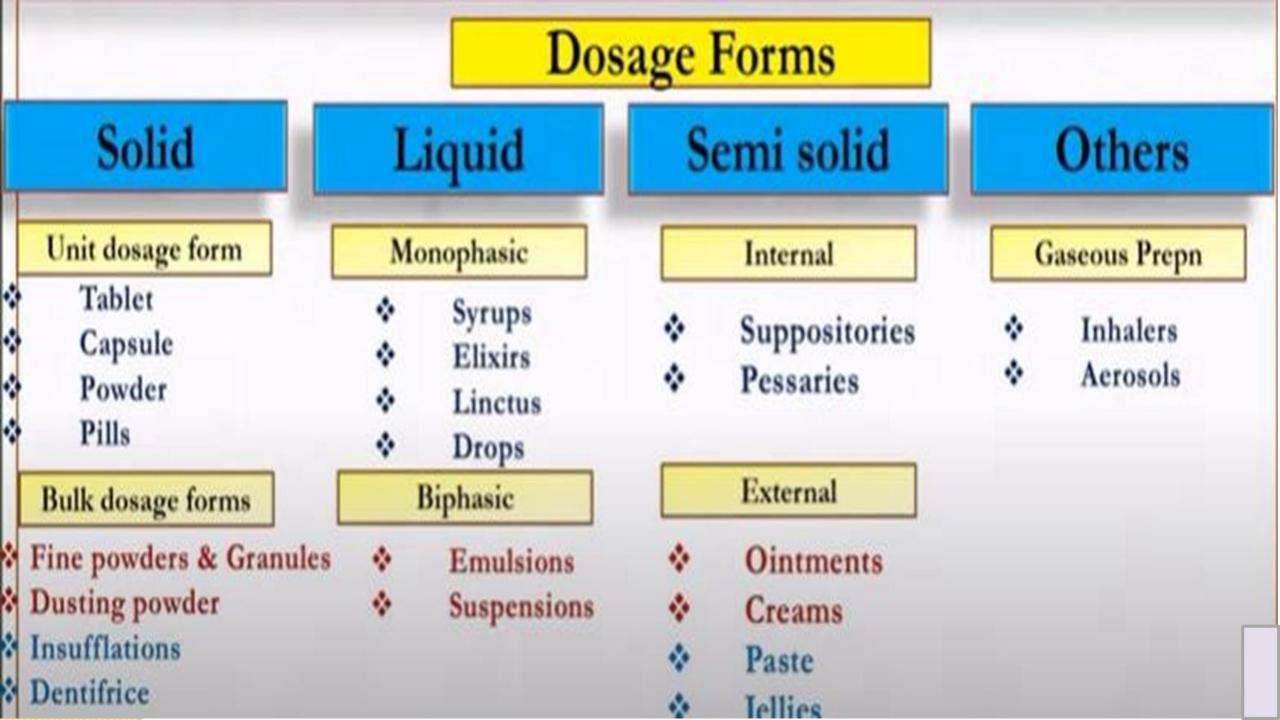


Dosage forms

By Assistant Lecture Abdulazeez .M.H

DOSAGE FORM

Dosage forms are pharmaceutical drug products in the form in which they are marketed for use, with a specific mixture of active ingredients and inactive components



1- SOLID PREPARATIONS

- A- Tablets
- **B- Capsule**
- C- Pill

TYPES OF TABLETS



TABLETS

Ordinary tablets

prepared by forcing the powdered drug into solid mass

the powdered contain the drug alone or the drug with a suitable diluent into a solid mass using a mechanical machine with optimal degree of compression.

e.g. paracetamol tab.

ORDINARY TABLETS





Diluent: it is an inert substance (pharmacologically inactive) used to increase the size of the powder in order to make compression of tablet easier.

Some very common diluents in tablets include starch, cellulose derivatives, and magnesium stearate

COATED TABLET

A solid disc of one or more pharmaceutic agents that is coated with sugar

or a flavoring to mask the taste .

Enteric-coated, meaning that it is coated with a substance that resists dissolution in the stomach but allows release of the medication in the intestine.

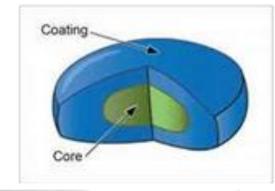
It has the following advantages :

1- to avoid the bitter taste of the drug

2-to prevent air oxidation of the drug
3- to facilitate swallowing in some patients
4- extending its shelf live. e.g. flu out tablet









ENTERIC COATED TABLE

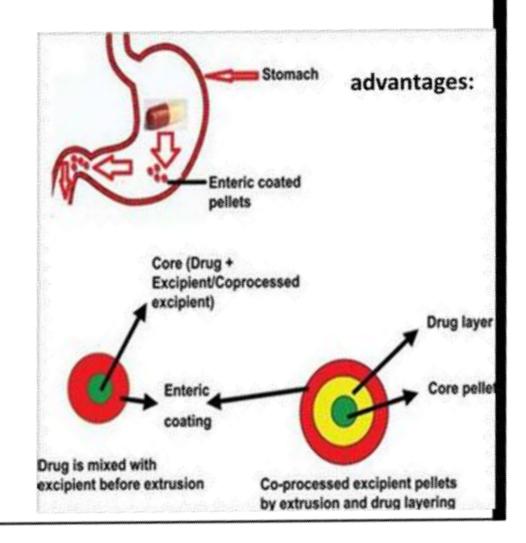
ordinary compressed tablet covered with acid resistant covering (salol) to allow the tablet to pass the stomach unchanged but is dissolved in the alkaline medium of the intestine

e.g. Aspirin

an oral dosage form in which a tablet is coated with a material to prevent or minimize dissolution in the stomach but allow dissolution in the small intestine..

ENTERIC COATED TABLET

- 1- avoid irritation
- 2- prevent drug destruction
- 3-to get local action



SUSTAIN RELEASE TABLET(SR)

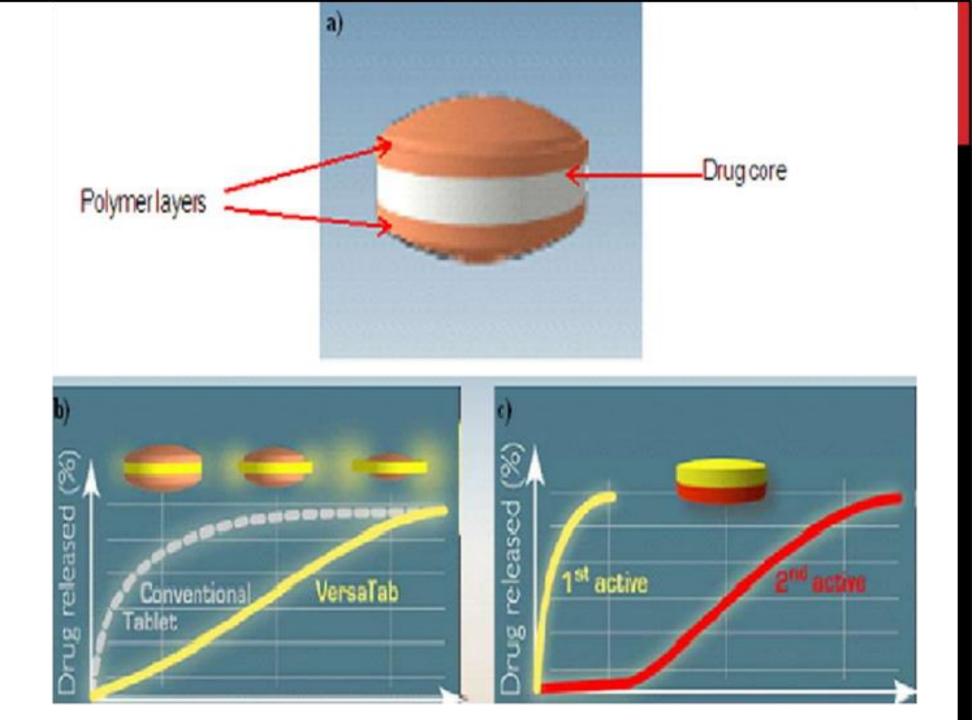
an ordinary tabdesigned to release (liberate) a drug at a predetermined rate in order to maintain a constant drug concentration for a specific period of time with minimum side effects..

Advantages:

1- prolong the duration of action

2-decrease the frequency of administration

e.g. glucophage retard tab.



SUSTAIN RELEASE TABLET







SUBLINGUAL TABLET

It is uncoated tab. especially manufactured to be suitable for absorption from sublingual mucosa Advantages rapid action avoid destruction by 1st pass metabolism e.g. glyceride trinitrates

SUBLINGUAL TABLET







EFFERVESCENT TABLET

Effervescent or carbon tablets are tablets which are designed to dissolve in water, and release carbon dioxide.

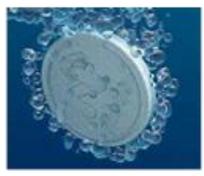
Large tab. Contain large dose ,manufactured by mixing the drug with citric acid and sodium bicarbonate to get granules.

The action of drug appears more rapid because the disintegration and dissolution takes place inside water and become ready for absorption .

EFFERVESCENT TABLET











CHEWABLE TABLET

Chewable tablets are an oral dosage form intended to be chewed and then swallowed by the patient rather than swallowed whole.

Ordinary uncoated tab. specially manufactured to be sucked or chewed

This tab usually with good taste

To bite and grind with the teeth; masticate.









LOZENGES

Sugar flavored tab with different shapes and attractive colors. It is sucked to treat tonsillitis and relief cough It contains volatile oils , antiseptic , antibiotics



LOZENGES









PASTILLES

They are solid medicated preparations designed to dissolve slowly in the mouth. They are softer than lozenges and their bases are either glycerol and gelatin, or and sugar





CAPSULES

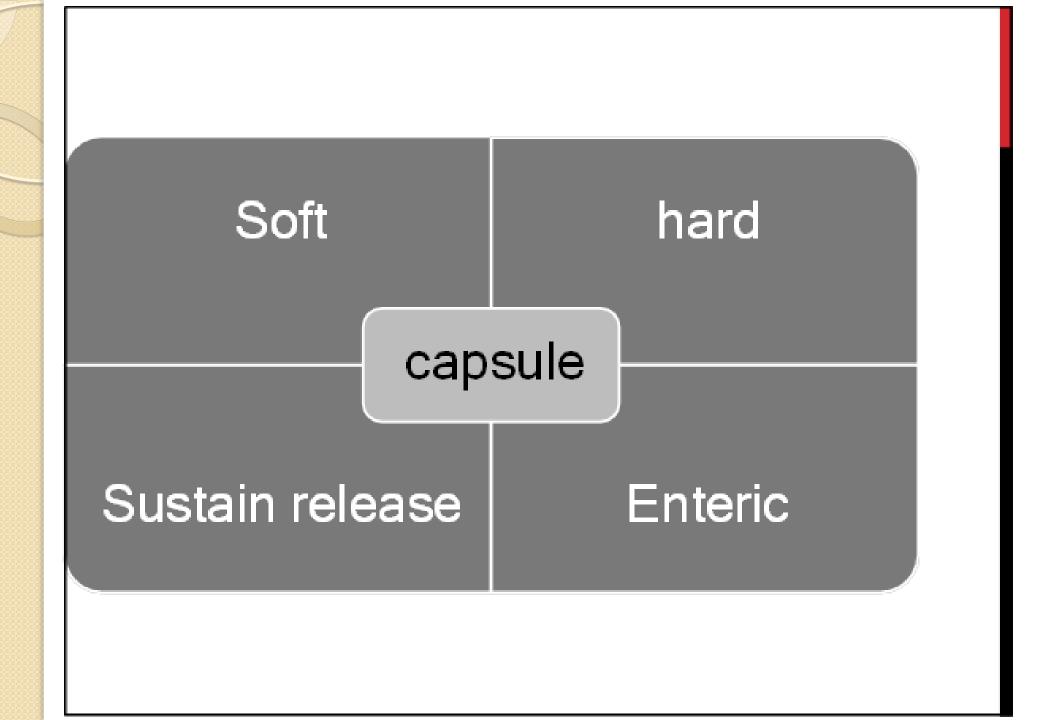
Ordinary capsules

Small cylindrical, oval, or rounded receptacles made of gelatin.

The two main types of capsules are:

1- Hard-shelled capsules, which are normally used for dry, powdered ingredients,

2- Soft-shelled capsules, primarily used for oils and for active ingredients that are dissolved or suspended in oil.



Cap . Shell which dissolve in the stomach is used for the following purposes

1- to mask the bad taste of the drug

2- to prevent air and moisture oxidation

3- to get accurate amount of the drug

ORDINARY CAPSULES





SPANCULE CAPSULE

Ordinary cap. Contains different granules each with different disintegration and dissolution rates

Advantages :

- 1- prolong the duration of action
- 2- decrease the frequency of administration

SPANCULE CAPSULE







SOFT CAPSULE





PILL

It is solid spherical body containing a drug in solid or liquid form given by mouth usually apill should not weight more than 0.3 g, it is sometime coated with sugar coat when the drug is liable to change during exposure to aim or when the drug has bitter taste

e.g. contraceptive

PILL







LIQUID PREPARATION

Syrup

It is a concentrated aqueous solution of a sugar, usually sucrose to which medicaments are added.

Flavored syrups are a convenient form to get good taste particularly for children .

SYRUP





Syrup of antibiotics supplied as powder form in which the antibiotic is mixed with specified quantity of sugar to be prepared as syrup by adding certain quantity of water, this is because antibiotics may hydrolyzed in aqueous solution to other component which either has no antibacterial activity or a substance causes allergic reactions, therefore all syrups of antibiotics must be discarded (7) days after adding water.

ELIXIR

It is a clear sweat - flavored liquid (usually contain alcohol)

It contains at least one active ingredient dissolved in a solution contains 15 – 50 % by volume of ethyl alcohol and is designed to be taken oral



EMULSION

It is amixture containg 2 immiscible liqiuds (such as oil and water)

One of which is broken up into minute globules.

Each globule is surrounded by a film of emulsifying agent and dispersed through out the other liquid .

Oil in water emulsion

Water serves as the dispersion medium Oil stays on top, acting as the dispersed phase

EMULSION





SUSPENSION

Liquid preparations for oral use containing one or more active ingredients suspended in a suitable vehicle.

- may show a sediment which is readily dispersed on shaking to give a uniform suspension which remains sufficiently stable to enable the correct dose to be delivered

SUSPENSION



POWDERED PREPARATIONS

1- ordinary powder : drug for internal use in form of fine powder mixed with water before administration

2- Effervescent powder active drug manufactured in form of effervescent granules by complexing the drug with sodium bicarbonate, citric acid, to be dissolved in water before ingestion.

EFFERVESCENT POWDER





DROPS FOR INTERNAL USE

This preparation is mostly convenient for infants, it is prepared by concentrating the drug in few drops to decrease volume of dose in order to facilitate swallowing of this small dose and minimize loss of the dose.



PARENTAL PREPARATION

Ampoule

It's a thin glass container for a single injectable dose the solution of ampoule is usually sterile indented to be use Im , sc , Iv

Ampoule for Iv injection usually contains very purified pyrogen free solution and most of Iv ampoule contain large volume in comparison with ampoule for Im injection

In some instances the active ingredient is putted in separated ampoule in form of powder and the solvent is putted in another separated ampoule to be mixed immediately before injection to avoid hydrolysis of the active ingredient

AMPOULE





VIALS

It is a thick glass container with rubber cap containing either solution or powdered drugs either for a single or multiple dose

VIALS









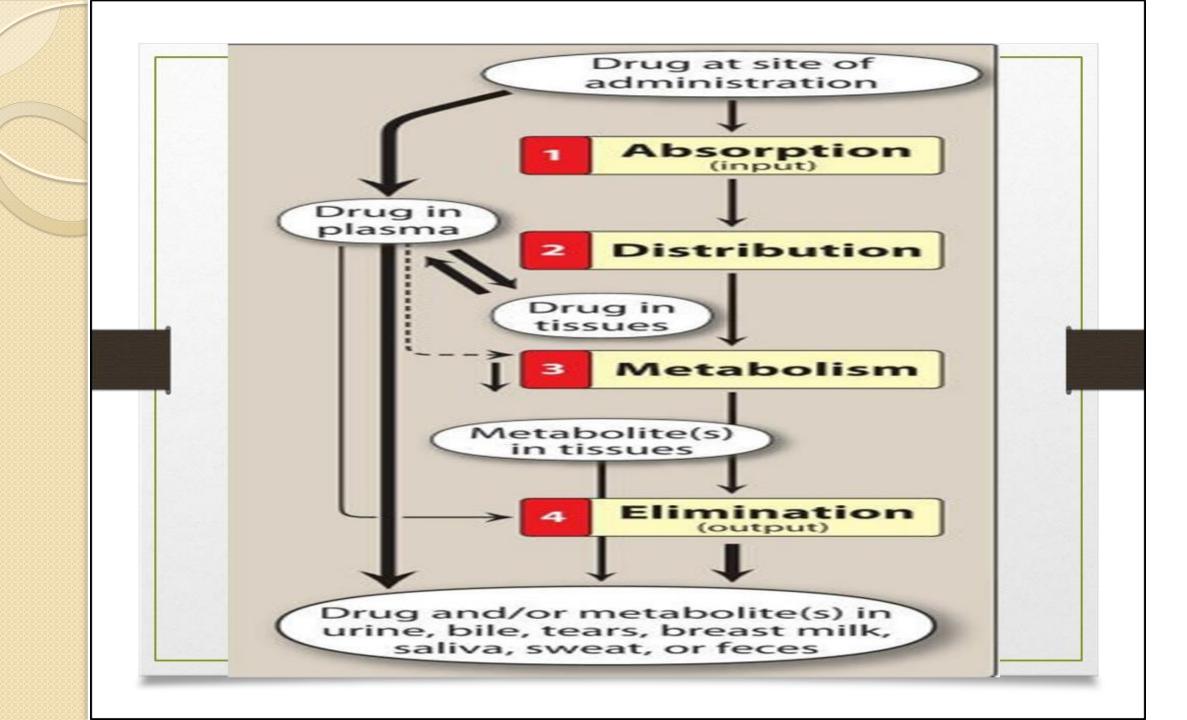
General Pharmacology

Lab Sessions



Saliva and Drug Excretion

By Assistant Lecture Abdulazeez .M.H



Major route of drug excretion

- Renal
- Liver
- intestine

Drug Clearance by the Kidney

Drugs must be sufficiently polar to be eliminated from the body.

Removal of drugs from the body occurs via a number of routes; the most important is elimination through the kidney into the urine.

Patients with renal dysfunction may be unable to excrete drugs and are at risk for drug accumulation and adverse effects.

Minor route of drug excretion

- Lung
- Bile
- Milk
- Saliva

Total body clearance

The total body (systemic) clearance, CL total, is the sum of all clearances from the drug-metabolizing and drugeliminating organs.

The kidney is often the major organ of excretion.

The liver also contributes to drug clearance through metabolism and/or excretion into the bile.

Total clearance is calculated using the following equation:

$$CL_{total} = CL_{hepatic} + CL_{renal} + CL_{pulmonary} + CL_{other}$$

Salivary secretion

 The drugs that secreted by saliva gain access to oral environment from the systemic circulation and can affect the microorganisms and tissue surfaces of the mouth.



Stimulated and Unstimulated Saliva

At rest, without exogenous or pharmacological stimulation, there is a small, continuous salivary flow, an unstimulated secretion, present in the form of a film that covers, moisturizes, and lubricates the oral tissues.

> This flow of saliva at rest is in the region of 0.4– 0.5mL/minute in healthy subjects.

Stimulated saliva is produced in response to a mechanical, gustatory, olfactory, or pharmacological stimulus, contributing to around 40-50% of daily salivary production.



- 1. passive diffusion across the cell of salivary gland.
- 2. passive diffusion across oral epithelium.
- 3. Flow of fluid from gingival cervices.

Example of salivary drug secretion

- Aspirin
- Phenytoin
- Ampicillin
- Diazepam
- Penicilin
- Tetracycline
- phenobarbital

• Most drugs are secreted in saliva are lipid soluble and enter from plasma to the saliva by simple passive diffusion.

- Some drug are secreted in gingival cervicular fluid at high conc. so it is useful in treatment of periodontal diseases like
- Tetracycline.

And other will causes side effects orally like

Gingival hyper plasia

e.g phenytoin,amelodipine.

And some which affect the normal flora like ampicillin.

Excretion of KI

- Iodine is essential for formation of thyroid hormons for synthesis of thyroxin.
- Deficiency of iodine thyroid hypertrophies

swelling of gland

- simple(non toxic goiter)
- Excessive intake of iodine
 - iodine mumps.

Uses of iodine

Treatment of goiter due to deficiency of iodine.
 Used as salageous agent to increase salivary secretion.
 It is used for sterilization.

4. Used as expectorant to encourage productive cough.

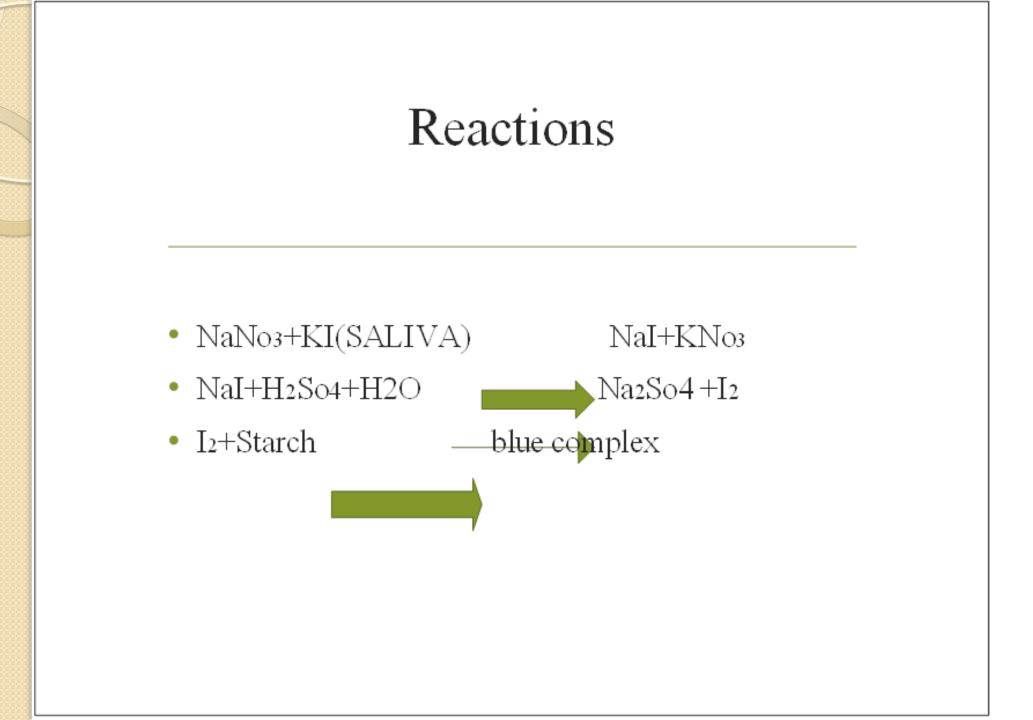
Procedure

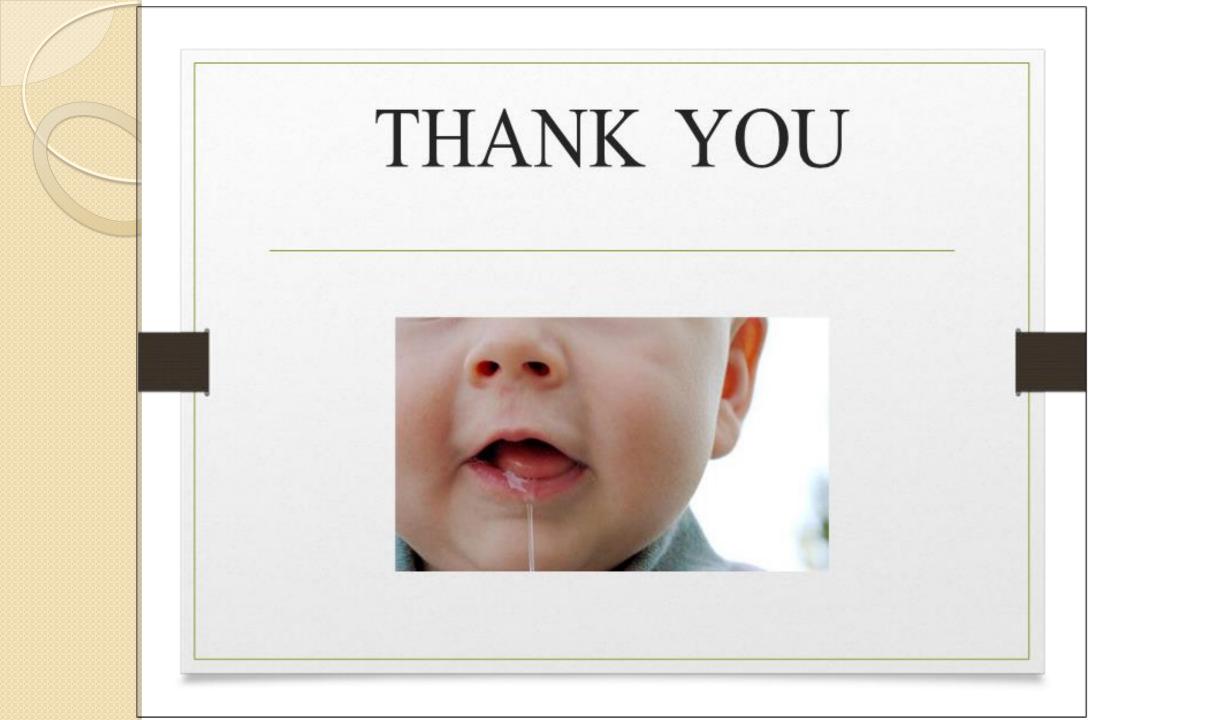
- Take 300 mg of KI in a capsule
- Take the saliva (unstimulated saliva) from volunteers.

- Add 1 ml of saliva in a test tube then add the following to it:
- 2-3 drops of 1% Sodium nitrite
- 2-3 drops 2N Sulphuric acid
- $1 \, \mathrm{ml} \, 1\%$ starch solution

- Test the samples of saliva for the presence of iodine at
- 5 min., 15 min. 30 min.

- Indicated the presence of iodine by using +,++,+++,++++.
- Arrange results in table.







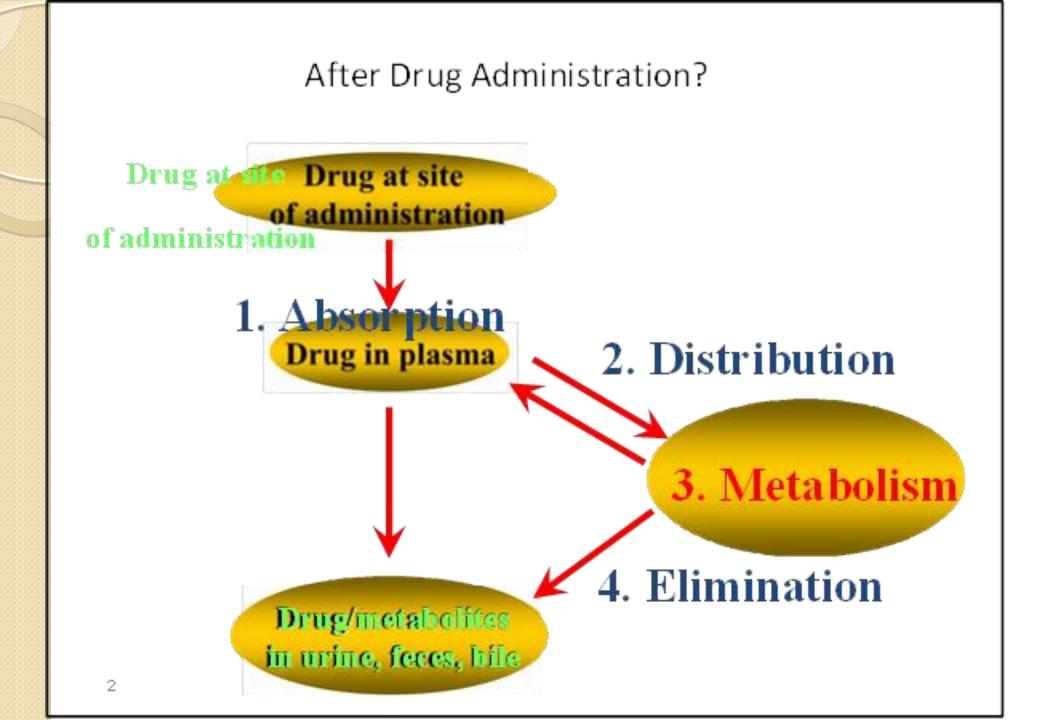
General Pharmacology

Lab Sessions



Effect of pH on drug absorption

By Assistant Lecture Abdulazeez .M.H



Factors Affecting Drug Absorption

- 1. Transport : active or passive
- 2. pH
- 3. Physical factors: include
 - blood flow
 - surface area
 - contact time

PH

- Drug are either weak acid or weak base
- Weak acid
- HA \longrightarrow H⁺+A⁻
- Weak base

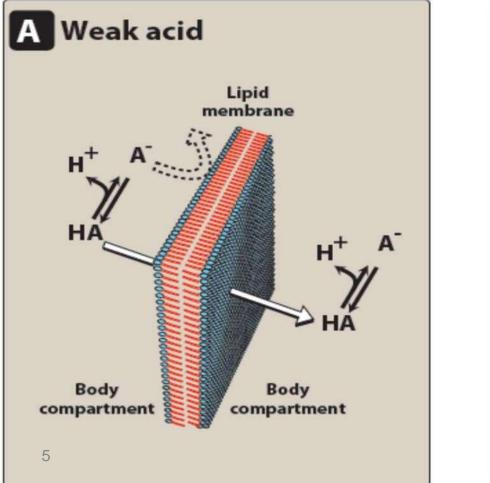
 $BH^+ \longrightarrow B+H^+$

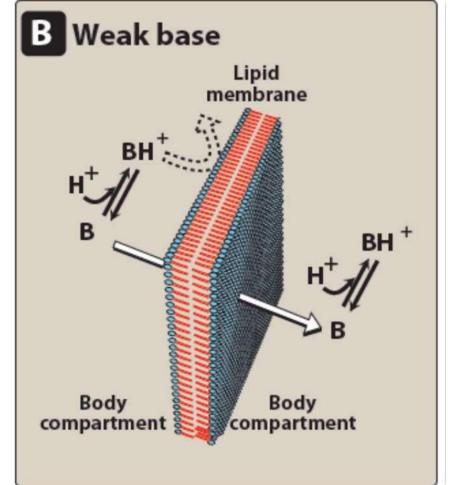
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pH of the medication

- The **pH** and **pKa** are the most important factors.
- The pH of the tissue determines the ratio of ionized to non-ionized drug. This ratio depends on the pKa of the drug.
- This amount will determine the easy with which the drug will penetrate or not through the tissues.

Drugs **pass** through ,membrane more **easy if it is uncharged (not ionized)** i.e ((HA) for acid) and ((B) for base) while charged **(ionized)** can **not passes**





the ratio between charged and

uncharged is determined by **PH** at site of absorption and by the **strength of acid a**nd base i.e **(pka)**

The **lower pka** of drug the **stronger the acid** ,conversely the **higher pka the stronger the base**

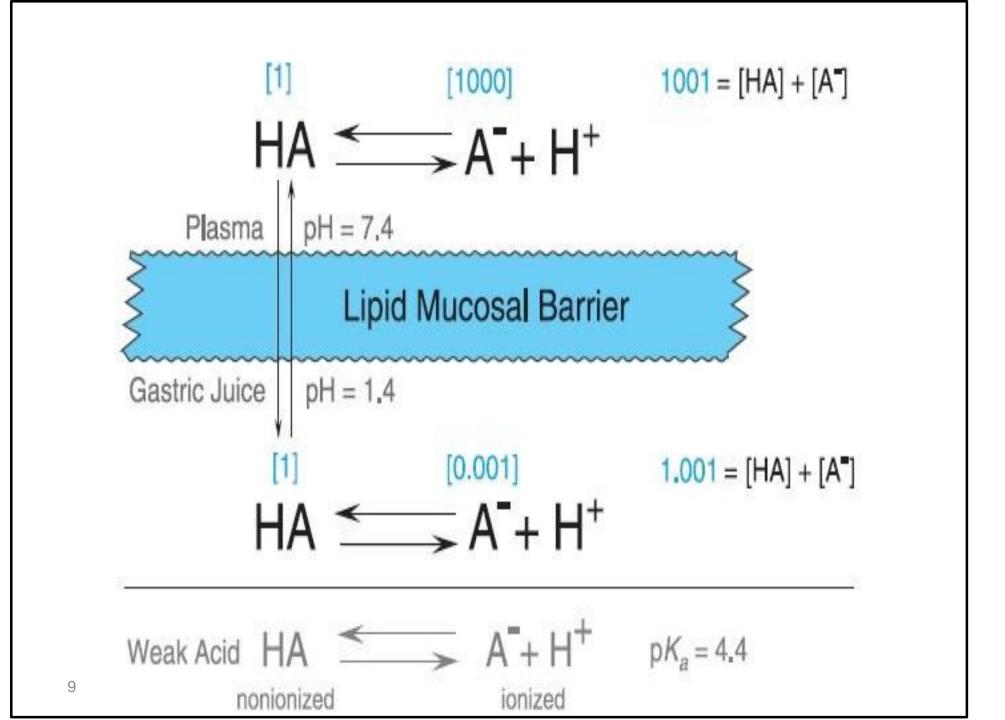
Henderson Hasselbalch Equation

- When pKa equals the pH where the ionized and nonionized forms are at equilibrium. (50% of each form is present.)
- the pKa pH relationship is described by the Henderson Hasselbalch equation, as follows:

For acid : pH - pKa = antilog [ionized / non-ionized]
For bases: pH - pKa = antilog [non-ionized /ionized]

- e.g. in case of infection or inflammation low
 natural PH (acidic tissue) cause less effect of L.A
 (weak base).
- This acidity results in a greater proportion of the ionized (charged) form of the anesthetic, thereby delaying or preventing the onset of action.
- i:e the PH < pKa (more drug in ionized form)

e.g. acidic drug (aspirin) unionized at acid gastric PH and absorbed from stomach.



Local anaesthetics are weak bases –

(Henderson-Hasselbalch equation)

```
pH - pKa = - log [non-ionized /ionized]
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<u>Example</u>: Calculate the proportions of free base and salt forms of drug X , (pKa = 8.5) at pH (7.5).

7.5 - 8.5 = - log unionized/ (ionized)

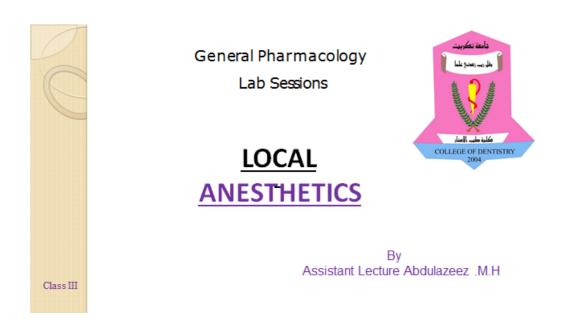
-1 = - log unionized/ (ionized)

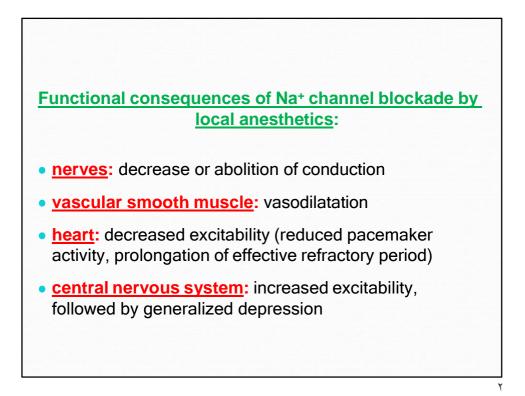
log 10 = log unionized/(ionized)

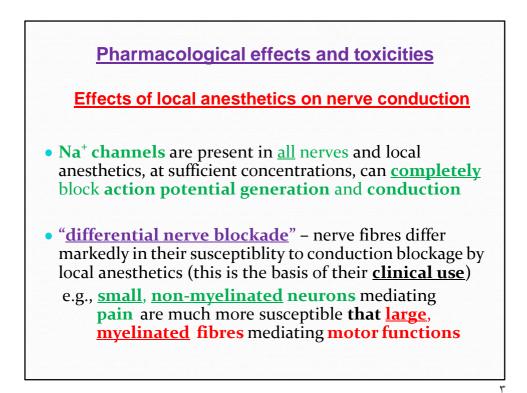
10/1 = unionized/ (ionized)

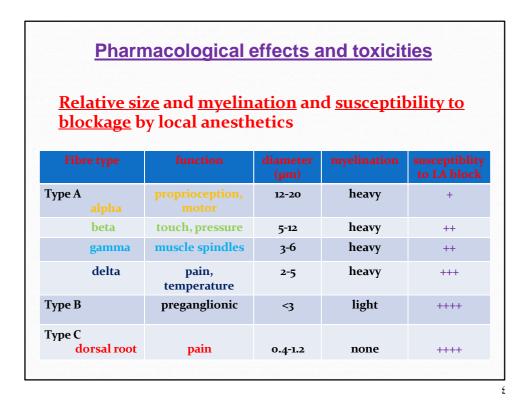
∴ there is 10x more drug in the ionized than (1) in the non-ionized form at physiological pH

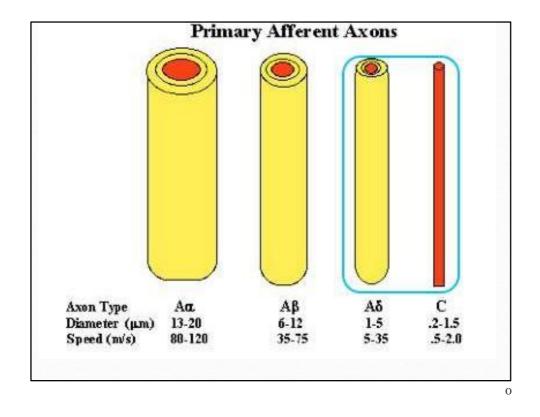


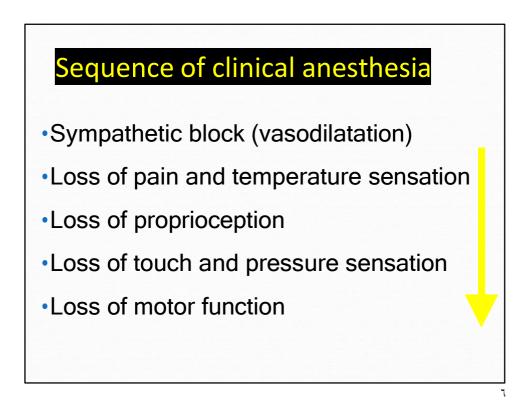


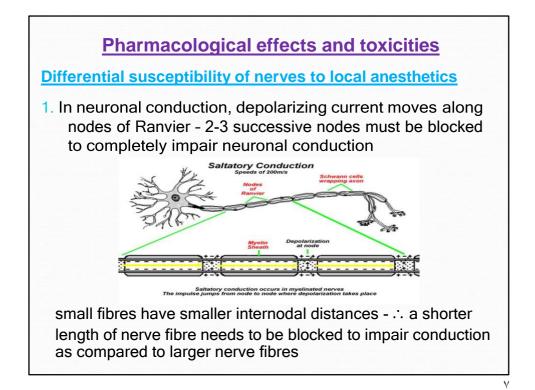


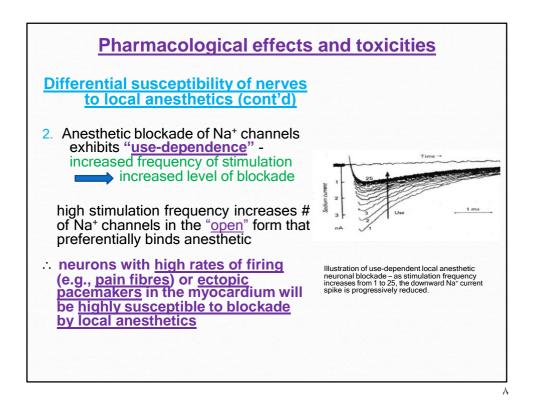


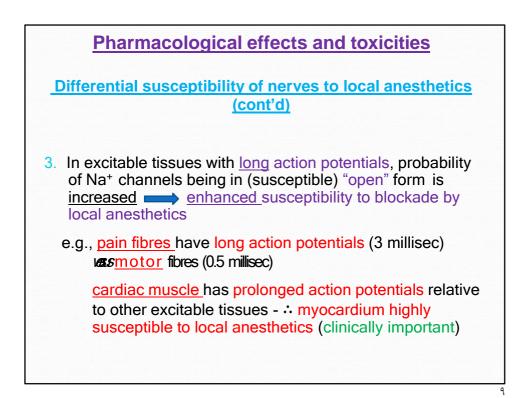


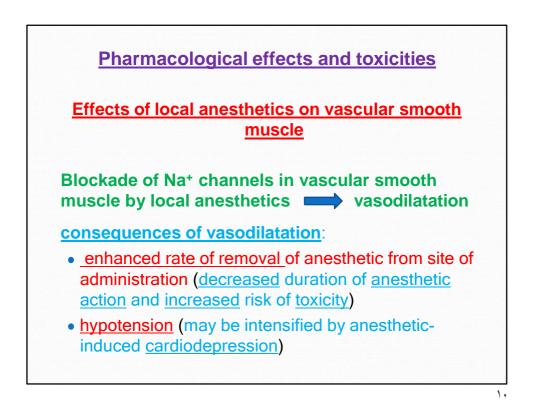


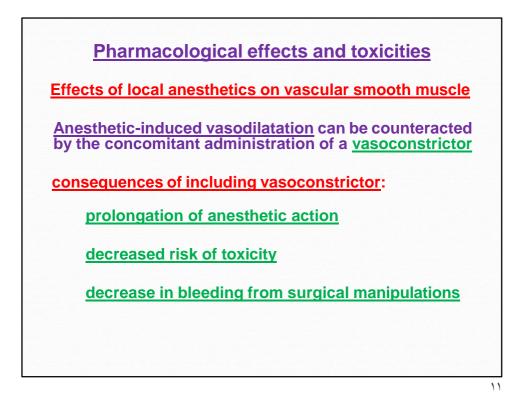












COMERCIALLY PREPARED LOCAL ANESTHESIA CONSISTS OF

In the construction of the constructio

Reducing agent (sodium metabisulphite)

Preservative (methylparaben,capryl

hydrocuprienotoxin)

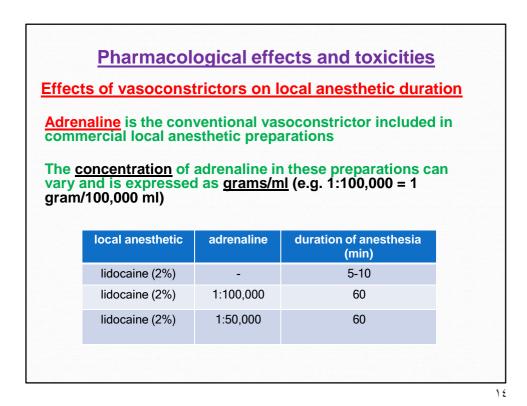
Pression Pression

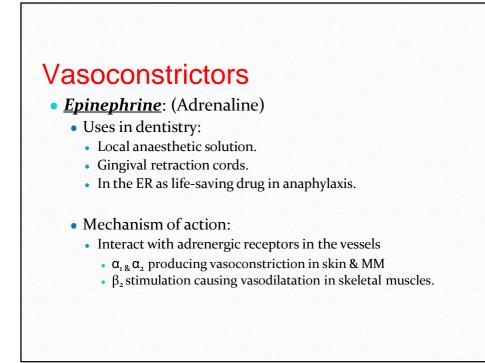
2Vehicle (distillde water,NaCl)

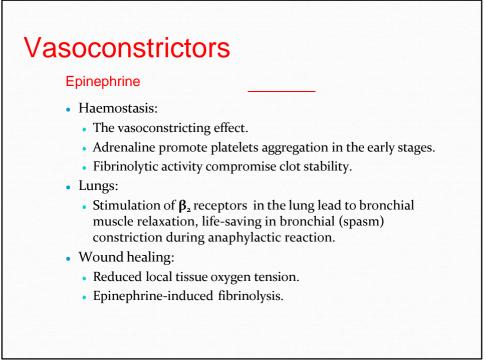
Vasoconstrictors

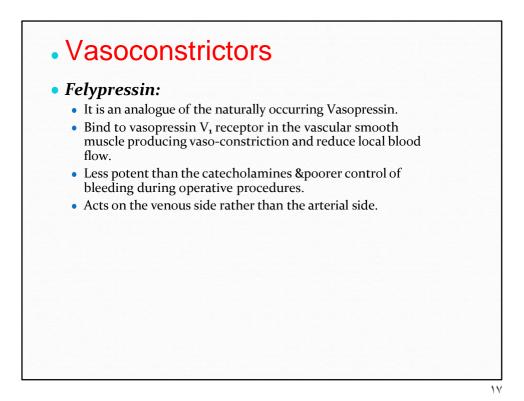
• Two types:

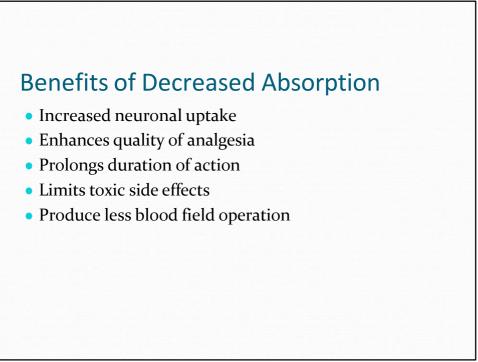
- Sympathomimetic naturally occurring.
- Synthetic polypeptides, Felypressin



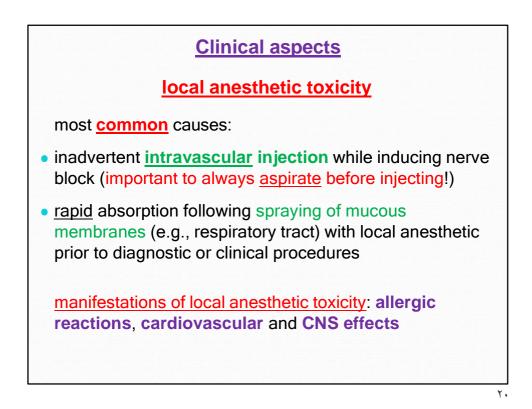


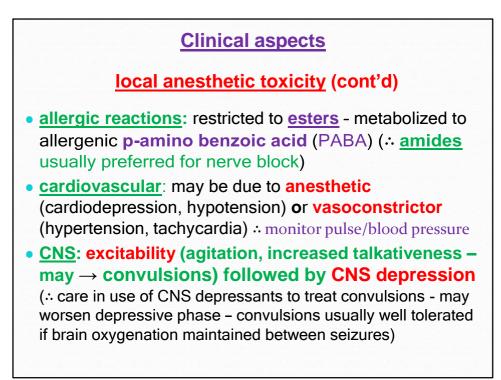




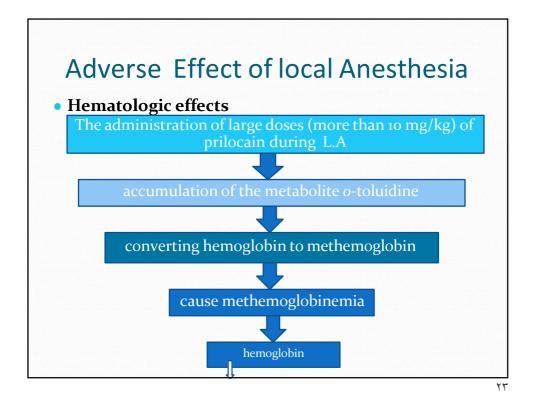


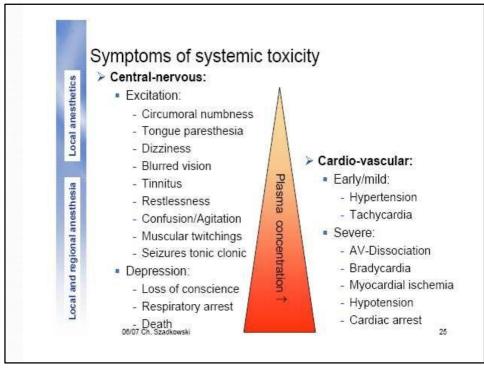
Тоу Мах	xicity ximum Dosa	literature nge (mg/lbs) X wei I Dosage (mg) ÷ m			
Ester	-		Maximum dose (mg/kg) vith Epinephrine	Duration of Action	lightheadedness timitus Purconsciousness
Ester	Procaine	7	1	0 60-90	Core a total and
	2-Choroprocaine	15		2030-60	perioral
	Tetracaine	1		.5 180-600	numbness
Amide				700.000	Toxicity Factors 1. Agent 2. Dowe 3. Administration late BP 0 1 4. Injection Sile writeror
	Lidocaine	4		790-200	A. Injection Site 5. +/- Vasoconstrictor 6. Acidity (pH)
	Mepivacaine	5		7 120-240	16. Acidaty
	Bupivacaine	2		3180-600	most toxic P B B B
	Levobupivacaine Ropivacaine	2		3180-600 3180-600	Only I can
	Articaine	2		760-230	KROSSand BRUCE.com
	Environne.			100.000	





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Treatment of toxicity

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• Treatment includes supportive measures. Excitement and convulsions may be controlled with 5 mg doses of diazepam or 2 mg doses of midazolam. Respiratory depression requires oxygen and possibly rescue breathing . hypotension Treatment includes patient positioning, IV fluids, and vasopressors. Cardiac asystole will require CPR.



General Pharmacology

Lab Sessions



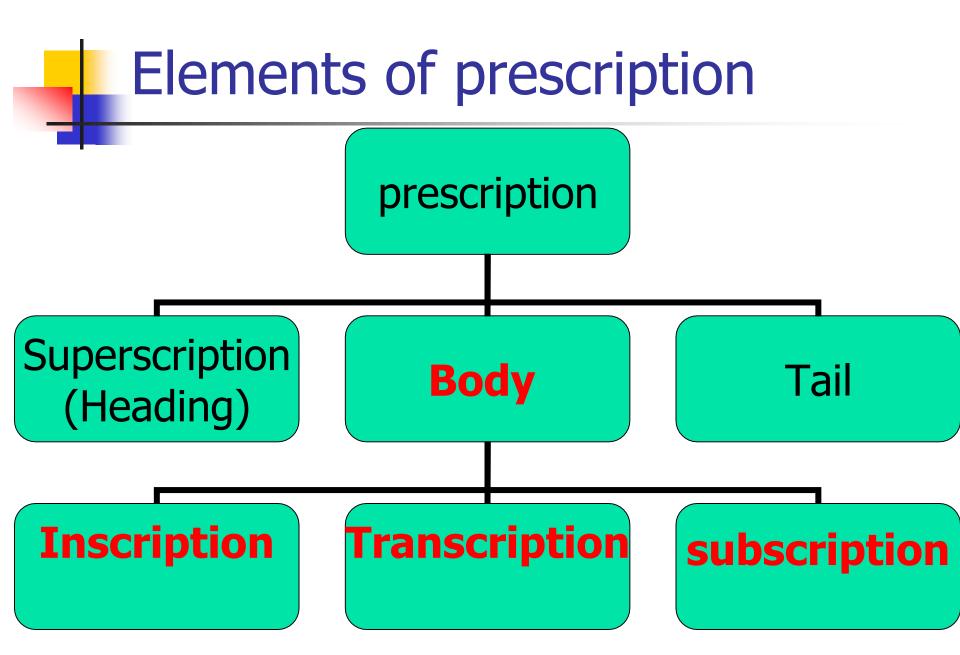


By Assistant Lecture Abdulazeez .M.H

Definition

Prescription

- Is term applied to formula written by physician/dentist
- to the pharmacist for the preparation of remedies.
- to the patient for the use of these drugs.



1. Superscription (Heading)

name sex age address name address teleph. no. qualification B.D.S, MSc.

Date Registration no. and date Diagnosis R (Recipe)

2. Body of prescription

- a. Inscription
- Name of drug
- Strength of drug
- Pharmaceutical form
- Amount of drug
- Tetracycline 250 mg capsule (20 caps)

2. Body of prescription

b. Subscription

Comprises directions from the physician to the pharmacist concerning the way of preparation to give the form and amount of prescribing drug. This is usually applied when there is compounded prescription.

Paracetamol 500 mg

- Diazepam 2 mg
- Mitte 10 caps

2. Body of prescription

c. Transcription

- Method of administration
- Amount to be taken
- Frequency of administration
- Time around meals
- Orally 1x4 before meal

Example (inscription and transcription):

Tetracycline 250 mg capsule (20 caps) Orally 1x4 before meal

3. Tail of prescription

- Refill directions.
- Type of bottle e.g antichildren bottle.
- Signiture.

- 1. Write in clear hand writing with correct names of drugs.
- 2. Date of prescription is important to detect cases where prescription orders are brought months or years after they were written by the prescriber.
- 3. Medicines are either prescribed using scientific or commercial names or both to avoid confusion with drugs having similar names.
- e.g. mefenamic acid. ponstane.

- 4. When the prescriber insist on certain brand produced by particular drug company write (please dispense as directed).
- 5. Write full name of prescriber.

6. Avoid personal abbreviations because it lead to misinterpretation. e.g.
Propranolol propoxyphene

- 7. Avoid as needed many times in the same prescription and write down exactly when the patient must use the drug.
- 8. Drugs and food interaction should be consider carefully to avoid decreasing activity of the drug.
- e.g. tetracycline and antacid.

- 9. Meals and drugs; generally :
- Drugs given 30 min before food (appetite stimulants, most antibiotics, all cholinergic, antiemetic).
- Drugs given within meals (drugs that cause stomach irritation (digestive enzymes).
- After food (NSAIDs).

Abbreviations

B.I.d twice daily
T.I.d three times daily
Q.I.d four times daily

a.c before mealP.c after meal

Compound prescription

- Drugs prepared by the pharmacist at the time of dispensing which are not available as patent preparation.
- They are prepared by mixing 2 or more drugs in a variety of dosage form including capsules, solutios, creams, ointments.

Compound prescription

Example:
Diazepam (2mg)
Paracetamol (500 mg)
Make (mitte) 10 cap.

Reasons for preparing Compound prescription

- To mask a well known drugs from the patient which the physician convinced that it is the required treatment e.g. diazepam, paracetamol.
- 2. Unavailability of small doses.
- 3. Unavailability of certain ingredients needed for treatment certain diseases.

